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Addition/Correction

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Correction to C–H Activation by a Diselenido Dinickel(II) Complex

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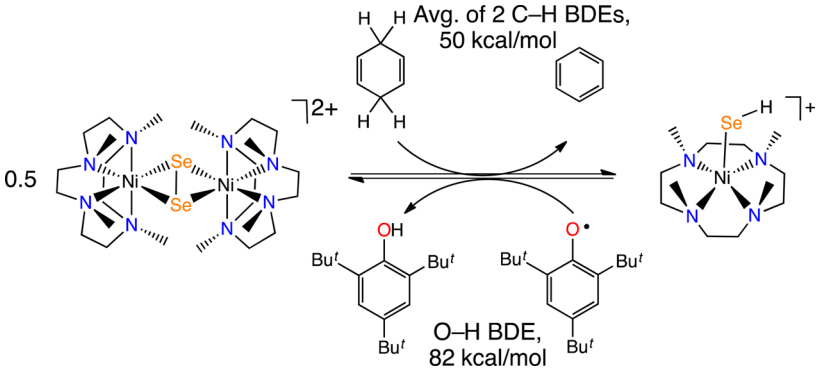
The thermodynamic analysis that led to the estimated Se−H effective bond dissociation enthalpy (BDE) of complex **2** considered only the BDE of removing the first H atom from 1,4-cyclohexadiene (~78 kcal/mol), when it should have taken into account the energies of removing both H atoms. When removing both the first and second H atoms is considered,[1] the lower limit for the Se−H effective BDE in **2** is ~50 kcal/mol. Accordingly, the Table of Contents artwork should be amended as follows:



REFERENCES

(1) Gao, Y.; DeYonker, N. J.; Garrett, E. C.; Wilson, A. K.; Cundari, T. R.; Marshall, P. *J. Phys. Chem. A* **2009**, *113*, 6955.

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